

A NEW ANTIBIOTIC, OKICENONE

II. PHYSICO-CHEMICAL PROPERTIES AND STRUCTURE ELUCIDATION

SHINJI FUNAYAMA[†], MASAMI ISHIBASHI^{††}, KANKI KOMIYAMA
and SATOSHI ŌMURA*

The Kitasato Institute, and School of Pharmaceutical Sciences of Kitasato University,
5-9-1 Shirokane, Minato-ku, Tokyo 108, Japan

(Received for publication March 1, 1991)

The structure of a new cytotoxic antibiotic, okicenone was elucidated to be 3,4-dihydro-4,6,9-trihydroxy-8-methyl-1(2*H*)-anthracenone on the basis of spectroscopic methods.

In the course of a screening program for novel antibiotics showing cytotoxic activity, okicenone (**1**) was isolated from the culture broth of *Streptomyces* sp. KO-3599 which had been isolated from a soil sample collected in Okinawa Prefecture, Japan. The isolation procedure and physico-chemical properties of the new antibiotic together with the taxonomic studies of the producing strain were reported in the preceding paper¹. This paper deals with the structure elucidation of okicenone (**1**).

Materials and Methods

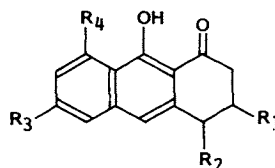
General Experimental Procedures

MP's were determined using a Yanagimoto MP-3 hot stage microscope and are uncorrected. UV spectra were recorded on a Shimadzu model UV-200S spectrophotometer. IR spectra were recorded on a Jasco model A-102 interferometer. MS were obtained with a Jeol model DX-300 mass spectrometer. ¹H and ¹³C NMR spectra were recorded on a Varian XL-400 instrument. Kieselgel 60 (Merck), Diaion HP-20 (Mitsubishi Chemical Industries) and Sephadex LH-20 (Pharmacia Fine Chemicals) were used for column chromatography and DC-Fertigplatten Kieselgel 60 (Merck) was used for TLC analysis and for preparative TLC.

Results and Discussion

Structure of Okicenone

Physico-chemical properties of okicenone (**1**) are summarized in Table 1. The UV and IR absorption spectra of **1** are shown in Figs. 1 and 2, respectively. The antibiotic is soluble in CHCl₃, EtOAc and MeOH but practically insoluble in H₂O. Okicenone gave a positive color reaction with iodine, 50% sulfuric acid and FeCl₃ and was negative to DRAGENDORFF's reagent and ninhydrin. The mo-



	R ₁	R ₂	R ₃	R ₄
1	H	OH	OH	CH ₃
2	OH	H	OH	CH ₃
3	H	OH	CH ₃	OH
4	OH	H	CH ₃	OH

[†] Present address: Pharmaceutical Institute, Tohoku University, Aobayama, Sendai 980, Japan.

^{††} Present address: Faculty of Pharmaceutical Sciences, Hokkaido University, Sapporo 060, Japan.

Table 1. Physico-chemical properties of okicenone (1).

Appearance	Pale yellow needles	IR ν_{\max} cm^{-1}	Fig. 2
MP	209~210°C	EI-MS (m/z)	258 (M^+), 240, 202, 201
Optical rotation		HREI-MS (m/z)	
$[\alpha]_D^{25}$	+20° (c 0.1, acetone)	Obsd	258.0879
TLC (silica gel)		Calcd for $C_{15}H_{14}O_4$	258.0892
$CHCl_3 - CH_3OH$ (15:1)	0.29	Color reaction	
$CHCl_3 - CH_3OH$ (9:1)	0.65	Positive	50% $H_2SO_4 + \Delta$, iodine, $FeCl_3$
Hexane-acetone (1:1)	0.42	Negative	Ninhydrin reagent, DRAGENDORFF's reagent
Molecular formula	$C_{15}H_{14}O_4$		
MW	258		
UV λ_{\max} nm	Fig.1		

Fig. 1. UV spectrum of okicenone (1).

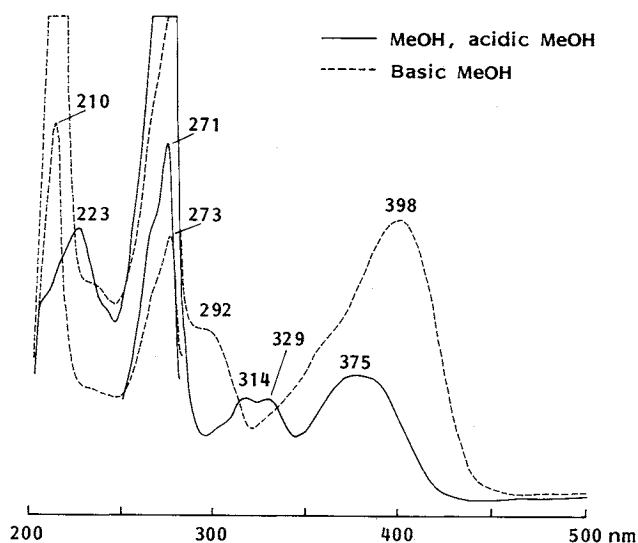
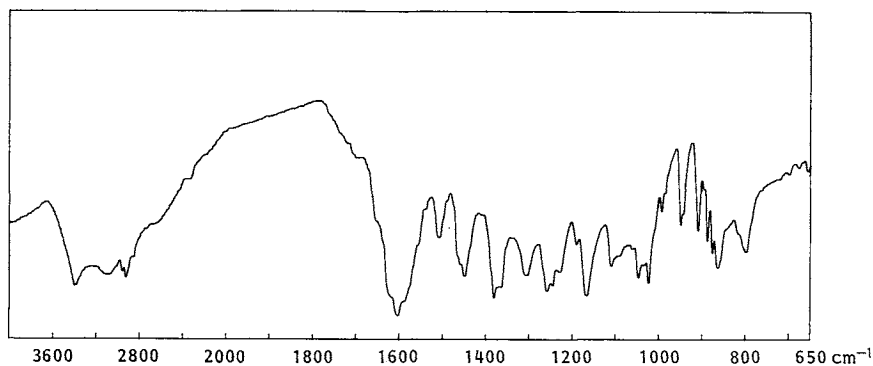


Fig. 2. IR spectrum of okicenone (1).

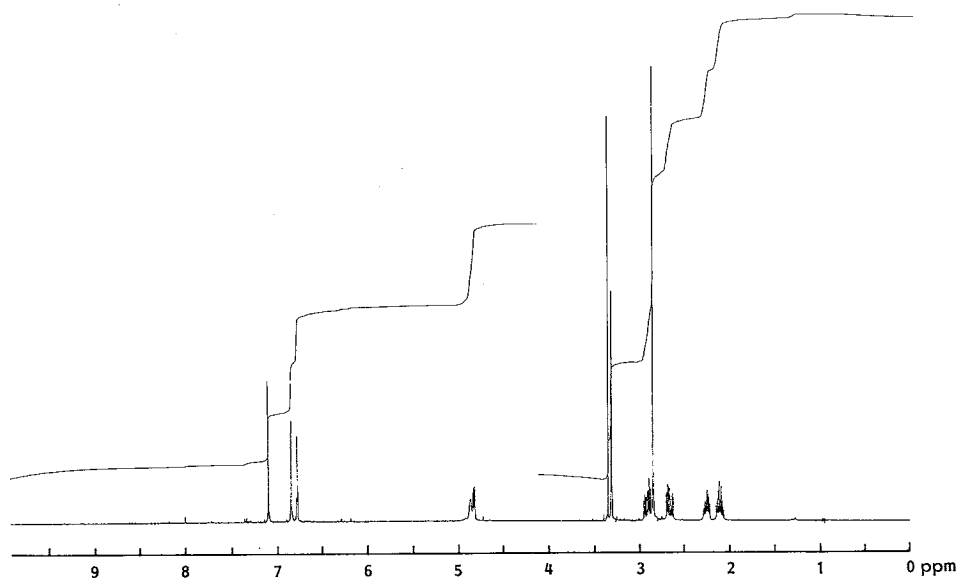


molecular formula of **1** was determined to be $C_{15}H_{14}O_4$ by HREI-MS (m/z 258.0879, $\Delta -1.3$ mmu). The UV spectrum (Fig. 1) showed absorption maxima at 223, 271 and 375 nm. These absorptions appeared similar to those of the aureolic acid group such as chromomycins²⁾, and indicative of an anthracenone moiety contained in the molecule of okicenone (**1**).

Table 2. ^1H and ^{13}C NMR chemical shifts of okicenone (**1**) in CD_3OD and protons to which an NOE was observed.

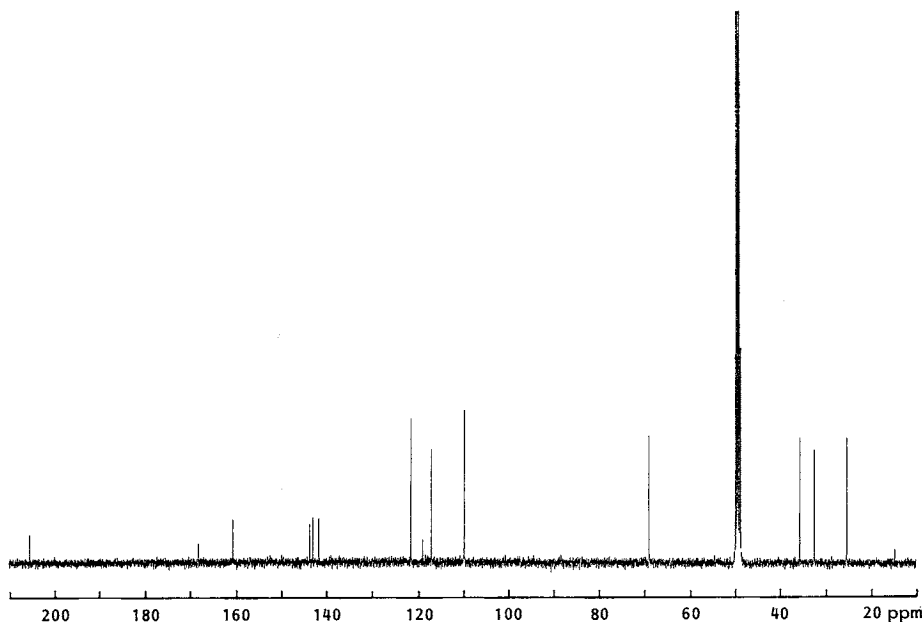
Position	^{13}C	^1H	NOE	Position	^{13}C	^1H	NOE
1	205.3 s			7	121.5 d	6.77 dq	8- CH_3 (2%)
2	35.7 t	(ax) 2.66 ddd (eq) 2.91 ddd		8	143.1 s		
3	32.5 t	(ax) 2.10 dddd (eq) 2.25 dddd		8- CH_3	25.4 q	2.84 d	7-H (14%)
4	68.9 d	4.83 ddd	10-H (6%)	8a	119.0 s		
4a	141.7 s			9	168.2 s		
5	109.7 d	6.84 d	10-H (10%)	9a	109.8 s		
6	160.7 s			10	117.0 d	7.10 d	5-H (13%)
				10a	143.8 s		

J (H,H) in Hz: $2_{\text{ax}},2_{\text{eq}}=18$; $2_{\text{ax}},3_{\text{ax}}=8$; $2_{\text{ax}},3_{\text{eq}}=5$; $2_{\text{eq}},3_{\text{ax}}=5$; $2_{\text{eq}},3_{\text{eq}}=8$; $3_{\text{ax}},3_{\text{eq}}=13$; $3_{\text{ax}},4=8$; $3_{\text{eq}},4=3.5$; $4,10=1$; $5,7=2.5$; $7,8\text{-CH}_3=1$.

Fig. 3. ^1H NMR spectrum of okicenone (**1**).

^1H and ^{13}C NMR spectra of **1** (Table 2, Figs. 3 and 4) along with DEPT experiments revealed the presence of one carbonyl, three sp^2 methines, seven sp^2 quaternary carbons, one sp^3 oxymethine, two sp^3 methylenes and one methyl group. This accounts for 15 carbons and 11 protons. The presence of three hydroxyl groups was inferred from the molecular formula $\text{C}_{15}\text{H}_{14}\text{O}_4$. The ^1H - ^1H COSY spectrum indicated a proton network: $\text{X}-\text{CH}_2-\text{CH}_2-\text{CH}(\text{OH})-\text{X}$. Since consideration of the unsaturation number implied that three rings were contained in the molecule of **1**, all the observations described above indicated a 3,4-dihydro-1(2H)-anthracenone nucleus with one methyl and two hydroxyl groups on the aromatic rings and a hydroxyl group on the C-2 or C-4 position. To determine the location of the hydroxyl and methyl groups, NOE experiments were carried out as follows (Fig. 5).

Upon irradiation of the sp^3 oxymethine proton at δ 4.83, a significant NOE (6%) was observed with the aromatic proton at δ 7.10, indicating the hydroxyl group at C-4 and the δ 7.10 proton at 10-H. Irradiation of the 10-H signal (δ 7.10) resulted in a 13% NOE in the aromatic proton at δ 6.84, which

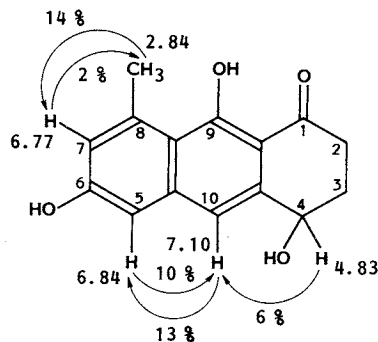
Fig. 4. ^{13}C NMR spectrum of okicenone (1).

was therefore assigned to 5-H. Converse irradiation of 5-H yielded a 10% NOE in the 10-H signal. Since the 5-H signal appeared as a doublet with a typical J -value (2.5 Hz) for *meta*-coupling, the third aromatic proton (δ 6.77) was assigned to 7-H; accordingly two hydroxyl and one methyl substituents had to be placed at the remaining C-6, C-8 and C-9 positions.

Irradiation of the methyl protons (δ 2.84) caused a strong NOE (14%) in the 7-H signal (δ 6.77) but an NOE was not observed in the 5-H signal (δ 6.84). This observation clearly showed that the methyl group was on C-8 and the hydroxyl group on C-6. The remaining hydroxyl group was therefore deduced to be on C-9. Additional proof for these results was provided by the study on the ^1H - ^{13}C long-range couplings through long range selective proton decoupling experiments to show the following connectivities: 4-H/C-4a, 4-H/C-10, 10-H/C-10a, 10-H/C-5, 5-H/C-10, 5-H/C-6, 7-H/C-6, 7-H/C-8, 8- CH_3 /C-7 and 8- CH_3 /C-8. Thus the structure of okicenone was concluded to be 3,4-dihydro-4,6,9-trihydroxy-8-methyl-1(2*H*)-anthracenone (1).

For those bearing structural similarity to okicenone (1), aloesaponols II (2)³, III (3)⁴ and germichryson (4)^{5~7} have been reported having the same molecular formula and very similar chemical properties.

Fig. 5. NOE experiments of okicenone (1).



Acknowledgment

This work was supported in part by Grants-in-Aid from the Ministry of Health and Welfare, the Ministry of

Education, Science and Culture, Japan, and by funds from Japan Keirin Association.

References

- 1) KOMIYAMA, K.; S. FUNAYAMA, Y. ANRAKU, M. ISHIBASHI, Y. TAKAHASHI, T. KAWAKAMI & S. ŌMURA: A new antibiotic, okicenone. I. Taxonomy, fermentation, isolation and biological characteristics. *J. Antibiotics* 44: 814~818 1991
- 2) UMEZAWA, H., (Ed.): "Index of Antibiotics from Actinomycetes", p. 215, Japan Antibiotics Res. Assoc., Tokyo, 1967
- 3) YAGI, A.; K. MAKINO & I. NISHIOKA: Studies on the constituents of *Aloe saponaria* HAW. I. The structures of tetrahydroanthracene derivatives and the related anthraquinones. *Chem. Pharm. Bull.* 22: 1159~1166, 1974
- 4) YAGI, A.; K. MAKINO & I. NISHIOKA: Studies on the constituents of *Aloe saponaria* HAW. II. The structures of tetrahydroanthracene derivatives, aloesaponol III and -IV. *Chem. Pharm. Bull.* 25: 1764~1770, 1977
- 5) TAKAHASHI, S.; M. TAKIDO, U. SANKAWA & S. SHIBATA: Germichryson, a hydroanthracene derivative from seedlings of *Cassia torosa*. *Phytochemistry* 15: 1295~1296, 1976
- 6) KO, K. S.; Y. EBIZUKA, H. NOGUCHI & U. SANKAWA: Production of secondary metabolites by hairy roots and regenerated plants transformed with Ri plasmids. *Chem. Pharm. Bull.* 36: 4217~4220, 1988
- 7) TAKITO, M. & S. KITANAKA (Taisho Pharm.): Extraction of tetrahydroanthracene derivatives as anticancer agents and pharmaceutical formulations containing them. *Jpn. Pat.* 207213 ('87), Sept. 11, 1987 [CA 110: 82476m, 1989]